

[10537/119]

PROCESS FOR COATING HOLLOW BODIES

The invention relates to a process for coating hollow bodies, in which a powder mixture comprising a metal donor powder, an inert filler powder and an activator powder is provided, the powder mixture is brought into contact with an inner surface, which is to be coated, of the body, e.g. comprising an Ni-, Co- or Fe-base alloy, and is heated.

The known processes for the diffusion coating of components made from heat-resistant alloys, such as Ni-, Co- or Fe-base alloys, include what are known as the powder pack processes. A process of this type is disclosed in US 3,667,985, in which the component surfaces to be coated are brought into contact with a donor powder comprising titanium and aluminum, to which an inert filler material and a halide activator are admixed, and is heated. US 3,958,047 has disclosed a powder pack process in which the metallic component is brought into contact with a donor powder which contains aluminum and chromium and is diffusion-coated by heating.

These processes are particularly suitable for coating the outer surfaces of metallic components, producing layer thicknesses of between 50 and 100  $\mu\text{m}$ . However, the processes have inherent drawbacks when coating internal surfaces, and consequently the internal layer thicknesses which are achievable with relatively complicated geometrical forms with narrow gaps, tight angles or undercuts are limited and inadequate, generally being below 30  $\mu\text{m}$ . A problem in this respect is that the donor powders have only a low ability to flow and therefore do not sufficiently fill the cavities. Moreover, after the coating, the donor powder can only be removed from the cavities with difficulty, and it is not possible to avoid leaving residues, and also the donor powder sinters to the surfaces.

5 The abovementioned drawbacks of the powder pack processes can  
in part be eliminated using what are known as gas diffusion  
coating processes. One process of this type is known from US  
4,148,275, in which a powder mixture which contains, for  
example, aluminum is arranged in a first chamber and the  
metallic components to be coated are arranged in a second  
chamber of a vessel. The coating gas is generated by heating  
the powder and, using a carrier gas, is deposited on the outer  
and inner surfaces of the components to be coated. However,  
10 the gas diffusion coating processes have the drawback that the  
devices for carrying out the process, such as for example for  
the forced guidance of the coating gases, are complex and  
expensive compared to those used for the powder pack  
processes. Furthermore, in this case too, the internal layer  
thicknesses which can be achieved are limited, since the  
coating gas or the donor metal gas is depleted on its route  
through the cavities of the component and a layer thickness  
gradient is produced along the length of the cavity. Since  
process conditions mean that the layer thickness of the outer  
coating is greater than that of the inner coating, the service  
life of the component is limited on account of the thinner  
internal coating.

25 US 4,208,453 has disclosed a process for the diffusion coating  
of the inner and outer surfaces of components, such as gas  
turbine blades, in which a powder mixture comprises 10% of  
chromium donor powder with a particle size of from 10 to 20  $\mu\text{m}$   
and 90% of alumina granules with a particle size of from 100  
to 300  $\mu\text{m}$ . In addition, a metal halide is added as activator.  
30 This disclosure does not deal with measures for increasing the  
layer thickness in cavities of complicated geometry.

DE 30 33 074 A1 discloses a process for the diffusion coating  
of the inner surface of cavities, in which a metallic  
35 workpiece can be coated with an aluminizing diffusion powder  
mixture comprising 15% of aluminum powder with a particle size

of 40  $\mu\text{m}$  and 85% of alumina powder with a particle size of approximately 200 to 300  $\mu\text{m}$ , as well as an  $\text{NH}_4\text{Cl}$  powder.

5 US 5,208,071 discloses a process for aluminizing a ferritic component with an alumina slurry, followed by heat treatment, the slurry comprising at least 10% by weight of chromium, at least 10% by weight of inert filler material, at least 12% by weight of water, a binder and a halogen activator, and finally the coated ferritic component is heat-treated. The process  
10 technology involved means that the use of a slurry differs significantly from a powder pack coating process.

GB 2 109 822 A has disclosed a metal diffusion process with which diffusion coatings can be produced more quickly than in  
15 the powder pack process, the coating powder being in loose form and being kept in contact with the component to be coated, in particular including with its internal surface, by mechanical means during the heating. The composition of the coating powder may comprise 10 to 60% of chromium powder, 0.1  
20 to 20% of chromium halide and alumina.

The problem on which the present invention is based is that of improving a powder pack process of the generic type described in the introduction in such a way that the layer thicknesses  
25 of the internal coating are sufficiently great even in the case of cavities with relatively complex geometries.

According to the invention, the solution to this problem is characterized in that the inert filler powder is provided with  
30 a mean particle size which is approximately the same as the mean particle size of the metal donor powder.

The advantage is that, when the particle sizes are selected in this way, it is possible to increase the specific density  
35 without agglomeration of the powder mixture, for example on account of an excessively high metal donor powder content. It is also ensured that there is no premature depletion of the

donor metal. A powder mixture of this type has good flow properties and, in tight corners, gains access to internal cavities which are to be coated. It is possible to coat hollow bodies, such as guide vanes and rotor blades of gas turbines made from heat-resistant Ni-, Co- or Fe-base alloys. Even in tight corners or recessed regions of the cavities, the layer thicknesses of the internal coating lie in the range from 50 to 110  $\mu\text{m}$ , therefore ensuring that the internal coating functions as an oxidation-resistant and corrosion-resistant layer.

In a preferred configuration, the metal donor powder and the inert filler powder are provided with a mean particle size of greater than 40  $\mu\text{m}$ , so that it is possible to achieve good permeation of the coating gas through the bed of the powder mixture.

The powder mixture is preferably provided with a metal donor powder content of 10 to 25% by weight, in order to prevent agglomeration of the powder mixture and to ensure good permeation through the bed.

Furthermore, it is expedient for an alloy with a donor metal content of 20 to 80% by weight to be provided as the metal donor powder, so that a sufficiently great layer thickness is ensured on account of the high donor metal content.

It may be advantageous for a mixture of an alloy with a donor metal content of 40 to 70% by weight and an alloy with a donor metal content of 30 to 50% by weight to be provided as the metal donor powder, so that the depletion of the metal donor in the two alloys takes place in steps, i.e. with a time delay.

The metal donor powder and the inert filler powder may be provided with a mean or average particle size of 150  $\mu\text{m}$ . A powder mixture of this type has good flow properties and fills

NY 380733 v 1

The cavity has a length of approximately 160 mm. Its inner surfaces are spaced apart at between 2 and 6 mm and converge at two opposite end sections. To coat the inner surfaces of the guide vanes, a powder mixture comprising approximately 20% by weight of metal donor powder and approximately 80% by weight of inert filler powder is provided. AlCr is selected as the metal donor powder, and Al<sub>2</sub>O<sub>3</sub> is selected as the inert filler powder. The melting point of AlCr is at least approximately 100°C higher than the coating temperature of approximately 800°C - 1200°C, so that there is no diffusion bonding of the metal particles to one another or agglomeration.

An activator powder forms approximately 3% by weight, the powder selected being AlF<sub>3</sub>, i.e. a halide compound. Another example of a suitable activator powder is CrCl<sub>3</sub>. A compound of this type has to have a low vapour pressure at the coating temperature, so that it is retained throughout the entire coating process. Moreover, a halide compound of the donor metal, in this case aluminum, is used, in order to avoid agglomeration as a result of a chemical reaction of the halogen with the donor metal.

The aluminum content, i.e. the metal donor content, in the metal donor powder is 50% by weight.

The powder mixture which has been prepared in this way is introduced into the cavity of the guide vanes for the purpose of coating the internal surfaces. The subsequent coating takes place at 1080°C with a holding time of 6 h, while the external coating, i.e. the coating of the outer surfaces of the guide vane, can take place simultaneously in a single-stage process using a conventional powder pack process or alternatively by means of a gas-diffusion coating process.

In the internal coating which is deposited in this way, the Al content in the layer is between 30 and 35% by weight.

The mean particle size of the inert filler powder is 100  $\mu\text{m}$ , and is significantly greater than the particle size of the metal donor powder, which is 60  $\mu\text{m}$ . The aluminum content, i.e. the metal donor content, in the metal donor powder is 50% by weight.

The powder mixture which has been prepared in this way is introduced into the cavity of the guide vanes for the purpose of coating the internal surfaces. The subsequent coating takes place at 1080°C with a holding time of 6 h, while the external coating, i.e. the coating of the outer surfaces of the guide vane, can take place simultaneously in a single-stage process using a conventional powder pack process or alternatively by means of a gas diffusion coating process.

In the internal coating which is deposited in this way, the Al content in the layer is between 30 and 35% by weight.

In a second example, an inert filler powder ( $\text{Al}_2\text{O}_3$ ) with a mean particle size of approximately 100  $\mu\text{m}$  is once again selected, forming approximately 80% by weight of the powder mixture. As activator powder,  $\text{AlF}_3$  forming approximately 3% by weight of a powder mixture is selected and admixed.

Unlike in Example 1, the metal donor powder, which forms approximately 20% by weight of the powder mixture, comprises two fractions. The first fraction is an alloy comprising AlCr, in which the aluminum content is 50% by weight. In the second fraction, the donor metal content, i.e. the aluminum content, is lower, being 30% by weight. This measure can be used to optimize the coating process in such a manner that first of all the fraction with the lower Al content is depleted, but the coating process is continued by the fraction with the higher Al content. In this way, it is possible to increase the ductility of the layers on the inner surfaces of the guide vane.

The Al content in the inner layers is 24 to 28% by weight. The inner layer thicknesses are between 65 and 105  $\mu\text{m}$  and are therefore significantly above the layer thicknesses which can be achieved with the conventional (powder pack) processes.

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In a third example, the hollow body is a hollow turbine guide vane of a gas turbine, which is provided with an oxidation-resistant and corrosion-resistant layer by means of a powder pack coating process. The elongate cavity is approximately 180 mm long. The inner surfaces are spaced apart at between 2 and 6 mm and converge at two opposite, longitudinal-side end sections. To coat the inner surface of the guide vane, a powder mixture comprising approximately 15% by weight of metal donor powder and just below 85% by weight of inert filler powder is provided. Depending on the particular application, the metal donor powder content may lie in the range from 10 to 25% by weight. The metal donor powder is AlCr, and the inert filler powder is  $\text{Al}_2\text{O}_3$ . The activator powder used is a halogen compound, such as  $\text{AlF}_3$ , forming approximately 3% by weight. The activator powder is therefore a halide compound of the donor metal Al.

The mean particle size of the inert filler powder is approximately equal to the mean particle size of the metal donor powder, being 150  $\mu\text{m}$ . The proportion of the donor metal Al in the metal donor powder, which is an alloy, is 50% by weight. The specific density of a powder pack mixture is high not because of a high metal donor powder content, but rather because of the selected particle size distribution. With this bed of the powder pack mixture, there is sufficient permeation by the coating gases emanating from the halide compound.

To coat the inner surface of the turbine guide vane, the powder mixture which has been prepared in this way is introduced into its cavity. At the selected particle size distribution of the inert filler powder and of the metal donor powder, the bed has good flow properties and gains access even

to the tight corners of the cavity. The subsequent coating takes place at 1080°C for a holding time of 6 h. It may be carried out at the same time as the external coating, i.e. the coating of the outer surface of the turbine guide vane, which  
5 may take place using a conventional powder pack process or also using a gas diffusion coating process. Generally, the coating is carried out on a plurality of turbine guide vanes simultaneously.

10 The Al content in the internal coating which has been deposited in this way is between 30 and 35% by weight and consequently in a highly advantageous range, i.e. there is, for example, no embrittlement of the layer.

15 Even in tight corners or recessed regions of the cavities, the layer thicknesses are in the range from 60 to 110  $\mu\text{m}$ , thus ensuring that the internal coating function has protection against oxidation and corrosion.  
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